

# EFFECT OF CONFORMATION AND CONFIGURATION OF ARABINOSE RESIDUES ON THE CIRCULAR DICHROISM OF C-GLYCOSYLFLAVONES

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**Key Word Index**—C-Glycosylflavones; sugar conformations; C-arabinosylflavones; circular dichroism.

**Abstract**—Examination of a variety of arabinose containing C-glycosylflavones has shown that the sign and intensity of the CD band at 250–275 nm (charge-transfer band) reflect not only the point of attachment of the sugar to the flavone but also depend upon the absolute and anomeric configuration, ring-size and ring-conformation in addition to the preferred rotameric conformation of the sugar about the C-aryl, C-1" bond. A change in stereochemistry of arabinose from the  $\alpha$  to  $\beta$  anomer resulted in sign inversion of the 250–275 nm CD band for 6-C-L-arabinosylflavones. Furthermore, a 6-C-arabinosylflavone containing  $\alpha$ -L-arabinose exhibited an oppositely signed charge-transfer CD band in comparison to one which contained  $\alpha$ -D-arabinose. 6,8-Di-C-glycosylflavones containing arabinose and glucose exhibited CD bands resulting from contributions due to both sugars, if the arabinose was not present as the  $\beta$ -pyranose form ( $^1C_4$  conformation).

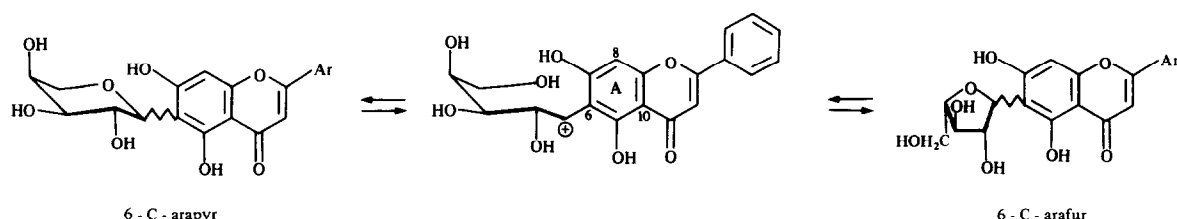
## INTRODUCTION

C-Glycosylflavones are widely distributed throughout the plant kingdom and have been isolated from sources as diverse as monocotyledons, dicotyledons, ferns, mosses and green algae [1]. While most of the C-glycosylflavones isolated in earlier reports contained only glucose, recent studies have described a variety of sugars attached to the flavone including arabinose. An interesting aspect of the chemistry of C-arabinosylflavones concerns their tendency to undergo an acid-catalysed isomerization of the carbohydrate leading to a mixture of anomeric C-arabinopyranosyl- and C-arabinofuranosylflavones which are readily separable on TLC after permethylation [2] (Scheme 1).

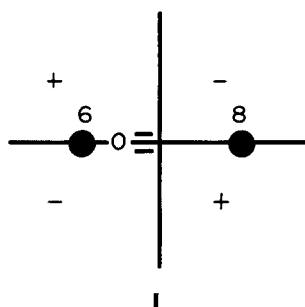
Although the structural determination of these compounds has primarily relied upon  $^1H$  NMR and MS data, CD has proved the configuration of L-arabinose in schaftoside and its isomers based upon identical curves shown by natural isoschaftoside and synthetic 6-C- $\alpha$ -L-arabinopyranosylvitexin [3] (*vide infra*). An earlier CD investigation of a wide range of C- $\beta$ -glycosylflavones showed that the sign of the 250–275 nm CD band is diagnostic of the point of attachment of the glycosyl residue to the phenolic moiety [4]. A positive CD band

at 250–275 nm indicated a 6-C-linkage, as in isovitexin (6-C- $\beta$ -D-glucopyranosylapigenin), whereas a negative CD band in this region signified an 8-C-linkage, as in vitexin (8-C- $\beta$ -D-glucopyranosylapigenin). 6,8-Di-C- $\beta$ -glycosylflavones generally showed two CD bands at 250–275 nm, a positive one at 263–275 nm and a negative one at 250–262 nm. The chiroptical properties of C- $\beta$ -glycosylflavones were explained by postulating a quadrant rule based upon the substituted benzoyl chromophore present in these molecules. Viewing the flavone from the CO end of the C-10, CO bond resulted in the sector diagram shown, (I) [4]. The sign of the CD band related to the charge-transfer transition (250–275 nm) should be determined by whichever sector or sectors are predominantly occupied by the asymmetric substituent or perturbing group which is the carbohydrate group in a C-glycosylflavone.

This study reports CD data on permethylated derivatives of mono 6-C-L- and 8-C-L-arabinosylflavones, free and derivatized 6,8-di-C-glycosylflavones containing L-arabinose and D-glucose, several 6,8-di-C-glycosylflavones containing L-arabinose and a sugar other than glucose and a 6-C-D-arabinosylflavone. In particular, the effect upon the CD of different anomeric configurations and ring sizes of the arabinose residue may assist in



Scheme 1.



the assignment of preferred conformations about the C-aryl, C-1" bond of the sugars in isomeric 6,8-di-C-glycosylflavones.

## RESULTS

### 6-C-L-Arabinosylflavones

CD spectra of perdeuteromethyl (PDM) ethers of isomeric 6-C-L-arabinosylacetins, 1–4, (Table 1) showed significant differences from each other (Fig. 1). The CD band at 250–275 nm of the  $\alpha$ -pyranosyl isomer, 1, corresponded in shape and appearance to that reported for isovitexin and its magnitude was similar to that shown by the permethyl and peracetyl derivatives of isovitexin [4]. The CD band at 262 nm of the  $\alpha$ -furanosyl isomer, 2, also resembled that reported for isovitexin but was only one-third as intense as that exhibited by 1.

The PDM ether of 6-C- $\beta$ -L-arabinofuranosylacetin, 3, gave a CD similar to that reported for free or derivatized vitexin but less than one-half as intense [4]. Clearly, changing the stereochemistry of the sugar from  $\alpha$  to  $\beta$  resulted in sign inversion of the charge-transfer band for 6-C-L-arabinofuranosylacetins. Different sector contributions (see I) of the  $\beta$ -sugar in comparison to those of the  $\alpha$ -sugar may reflect whether C-2" is positioned above or below the plane of the flavone A-ring (*vide infra*). Examination of the CD of the  $\beta$ -pyranosyl isomer, 4, yielded results less striking yet consistent with those shown by 3. Although the 250–275 nm region of 4 was relatively featureless, a weak negative CD band was present at 265 nm in agreement with  $\beta$ -stereochemistry.

### 6-C-D-Arabinosylflavones

In order to ascertain the effect of different absolute configurations of arabinose upon the CD of a C-glycosylflavone, 6-C- $\alpha$ -D-arabinopyranosylapigenin (5) was prepared and its CD measured (Table 1). The presence of a CD band at 264 nm, in 5, of opposite sign to that shown by the  $\alpha$ -L-compound [4] demonstrated that knowledge of the anomeric configuration of the arabinose residue from  $^1\text{H}$  NMR data will permit CD measurements to differentiate D-arabinose from L-arabinose in 6-C-arabinosylflavones.

### 8-C-L-Arabinosylflavones

The PDM derivative of 8-C- $\alpha$ -L-arabinopyranosylgenkwanin (PDM-6) showed a broad negative CD band at 250–275 nm (Fig. 2) approximately one-half as intense

as that recorded by its underivatized parent molludistine, 6, and by vitexin [4]. The high wavelength CD of the  $\alpha$ -furanosyl isomer, 7, resembles that shown by the  $\alpha$ -pyranosyl compounds, 6 and PDM-6, but 250–275 nm CD band, although of the same sign in the isomeric  $\alpha$ -anomers, is of much lower magnitude in 7 as noted in the previous section for related compounds of the 6-C-L-arabinosylflavone series. Acid isomerization of 8-C-arabinosylgenkwanin did not produce the  $\beta$ -pyranose and  $\beta$ -furanose isomers in quantities large enough to permit their isolation [2].

### 6,8-Di-C-glycosylflavones containing L-arabinose and D-glucose

Schaftoside, 8, isolated from *Silene schafta* in 1967 [5], was shown to be 6-C- $\beta$ -D-glucopyranosyl-8-C- $\alpha$ -L-arabinopyranosylapigenin on the basis of oxidative degradation and spectroscopic investigations [3] whereas its acid isomerization product, isoschaftoside (9), is identical with synthetic 6-C- $\alpha$ -L-arabinopyranosyl-8-C- $\beta$ -D-glucopyranosylapigenin [3]. The arabinosyl residue of both 8 and 9 gave rise to larger CD bands at 250–275 nm than the glucosyl residue [4]. While the negative lower wavelength band dominated the CD of the 8-C-arabinosyl compound, 8, the positive higher wavelength band was most prominent in the CD of the 6-C-arabinosyl isomer. Derivatized samples of 8 and 9 yielded less complex CD spectra, in the 250–275 region, than those of the parent compounds. The CD of PDM-8 gave only a positive band at 261 nm attributed to the 6-C-glucosyl substituent whereas the CD of PDM-9 exhibited no discernible bands at 250–275 nm.

Acid isomerization of schaftoside produced two additional isomeric 6,8-di-C-glycosylapigenins containing arabinose and glucose, neoschaftoside (10) [6] and neoisoschaftoside (11) [7], which have been found to be identical with schaftoside and isoschaftoside, respectively, except that the arabinose of 10 and 11 is present as the  $\beta$ -L-pyranosyl configuration [8] [Besson, E. and Chopin, J., unpublished results]. The CD of PDM-10, which was clearly dominated by the presence of the 6-C-glucosyl group, resembled that reported for the permethyl and peracetyl derivatives of isovitexin [4] whereas the underivatized flavone glycoside, 10, exhibited a weak positive CD band at 255 nm (Fig. 3).

The PM (permethyl) derivative of neoisoschaftoside (PM-11) afforded a featureless CD from 310 to 245 nm (Fig. 3). Synthetic PDM 6-C- $\beta$ -L-arabinofuranosyl-8-C- $\beta$ -D-glucopyranosylacetin (PDM-12) gave a CD similar to that of PM-11 at high and low wavelength but quite divergent at 250–275 nm reflecting the different contributions of a  $\beta$ -furanosyl and a  $\beta$ -pyranosyl sugar to the charge-transfer transition. The CD of underivatized neoisoschaftoside, 11, which is dominated by a strong negative band at 271 nm attributed to the 8-C-glucosyl group, provides further evidence of the cancellation of CD contributions due to  $\beta$ -arabinopyranosyl residues (c.f. 4, 10, PDM-10, PM-11).

### 6,8-Di-C-glycosylflavones containing L-arabinose and a sugar other than glucose

The CD of corymboside (6-C- $\alpha$ -L-arabinopyranosyl-8-C- $\beta$ -D-galactopyranosylapigenin), 13, resembled that shown by isoschaftoside (9) whereas the CD of compound

Table 1. Circular dichroism\* of C-arabinosylflavones,  $[\theta]$  ( $\lambda$ )

Compound	$[\theta]_{>275}$	$[\theta]_{<250}$	$[\theta]_{250-275}$
<b>6-C-L-Arabinosylacetins</b>			
PDM $\alpha$ -pyranosyl (1)	8080 (340)		
	-16,100 (310)	—	23,100 (262)
PDM $\alpha$ -furanosyl (2)	1870 (345)		
	-5140 (312)	-5420 (243)	7760 (262)
PDM $\beta$ -furanosyl (3)	1790 (340)	860 (240)	-6970 (266)
PDM $\beta$ -pyranosyl (4)	880 (350)		
	940 (335)	—	-1760 (265)
	-5430 (318)		
	420 (290)		
<b>6-C-D-Arabinosylapigenin</b>			
$\alpha$ -Pyranosyl (5)	5250 (395)	—	-19,700 (264)
	3060 (317)		
	-8970 (298)		
	21,700 (283)		
<b>8-C-L-Arabinosylgenkwansins</b>			
$\alpha$ -Pyranosyl (6)	4660 (350)		
	-8320 (282)	-7650 (238)	-12,000 (267)
PDM $\alpha$ -pyranosyl (PDM-6)	3510 (335)		-5800 (265)
	-3400 (285)	—	-5690 (253)
PDM $\alpha$ -furanosyl (7)	2590 (325)		
	-2220 (277)	—	-1930 (260)
<b>6,8-Di-C-glycosylflavones containing L-arabinose and D-glucose</b>			
Schaftoside (8)	12,500 (322)		2540 (275)
	17,100 (303)	—	-16,900 (250)
PDM-schaftoside (PDM-8)	10,300 (340)		
	-6670 (310)	-17,400 (235)	4040 (261)
Isoschaftoside (9)	3570 (345)		17,300 (272)
	-700 (322)	—	-5650sh (250)
	9030 (301)		
PDM Isoschaftoside (PDM-9)	20,200 (340)		
	-10,600 (307)	—	†
Neoschaftoside (10)	-3600 (318)	-6900 (216)	1160 (255)
PDM-neoschaftoside (PDM-10)	19,000 (345)		
	-11,700 (310)	-13,000 (230)	26,700 (262)
Neoisoschaftoside (11)	4340 (320)	-10,400 (233)	
	3470 (305)	15,600 (222)	-23,700 (271)
PM-neoisoschaftoside (PM-11)	4830 (337)	-7360 (234)	†
PDM-6-C- $\beta$ -L-arabinofuranosyl-			-5120 (265)
8-C- $\beta$ -D-glucopyranosylacetin (PDM-12)	4270 (340)	-6220 (235)	-5610 (250)
<b>6,8-Di-C-glycosylflavones containing L-arabinose and a sugar other than D-glucose</b>			
Corymboside (13)	3450 (379)		10,300 (274)
	-1410 (324)	-9100 (233)	-5950 (250)
	4930 (301)		
Compound S (14)	2360 (358)		5140 (271)
	-820 (327)	—	-4930sh (250)
6-C- $\beta$ -D-Xylopyranosyl-8-C- $\alpha$ -L-	5140 (303)		
arabinopyranosylapigenin (15)	11,100 (302)	—	-9870 (250)

\*CD values are expressed in terms of molecular ellipticity  $[\theta]$  in units of  $\text{deg}\cdot\text{cm}^2\cdot\text{dmol}^{-1}$  with wavelength ( $\lambda$ ) in nm. Measurements were obtained in methanol solutions at 27°.

†No maxima observed between 250 and 275 nm.

S (6-C- $\beta$ -D-galactopyranosyl-8-C- $\alpha$ -L-arabinopyranosyl-apigenin), 14, was similar to that of schaftoside, 8, with the 6-C-galactosyl group providing a greater contribution to the CD than a 6-C-glucosyl group as noted earlier for related compounds [4]. 6-C-Xylosyl-8-C-arabinosyl-apigenin, 15, gave a negative CD band at 250 nm presumably due to the 8-C-arabinose. Earlier studies have shown that the xylosyl group resulted in relatively weak CD

contributions in C-glycosylflavones due to loss of the hydroxymethyl group at C-5'' [4].

## DISCUSSION

### Conformational preferences of C-glycosylflavones

Interpretation of the CD spectra of C-glycosylflavones

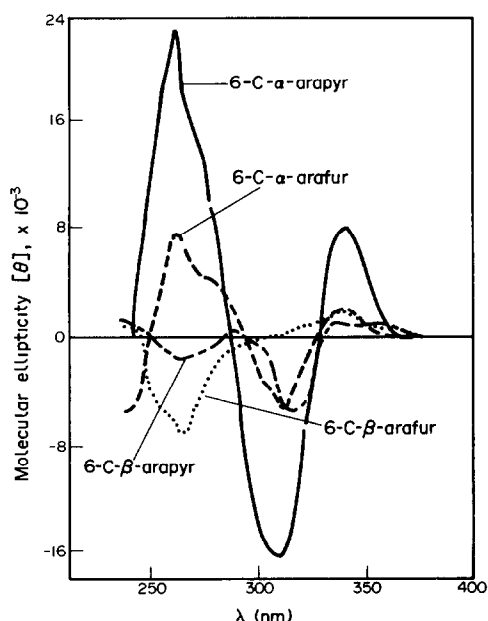


Fig. 1. Effect of different conformations and configurations upon the CD of derivatized 6-C-L-arabinosylflavones. CD spectra of PDM-6-C- $\alpha$ -L-arabinopyranosylacetin (1) (—), PDM-6-C- $\alpha$ -L-arabinofuranosylacetin (2) (---), PDM-6-C- $\beta$ -L-arabinofuranosylacetin (3) (···), PDM-6-C- $\beta$ -L-arabinopyranosylacetin (4) (—·—).

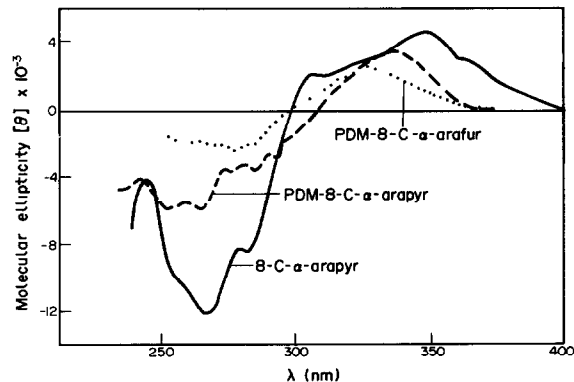


Fig. 2. Effect of different conformations and derivatization upon the CD of 8-C-L-arabinosylflavones. CD spectra of 8-C- $\alpha$ -L-arabinopyranosylgenkwanin (6) (—), PDM-8-C- $\alpha$ -L-arabinopyranosylgenkwanin (PDM-6) (---), PDM-8-C- $\alpha$ -L-arabinofuranosylgenkwanin (7) (···).

requires knowledge of both the preferred rotational conformation of the carbohydrate about the C-6 or C-8, C-1" bond and the ring conformation of the sugar heterocycle. Evidence for the former suggests that C-

\*When C-1" is below the reference plane of the pyranoid ring (the plane containing 0, C-2", C-3" and C-4"), the chair conformer is designated as  $^4C_1$ ; when it is above the reference plane, as  $^1C_4$  [13].

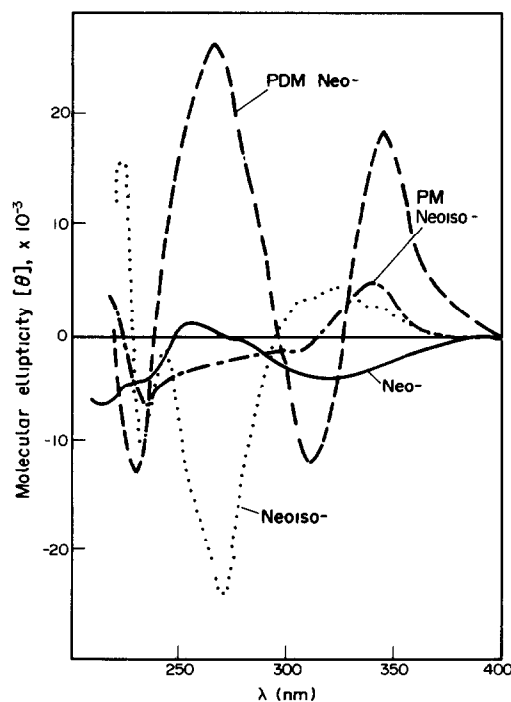
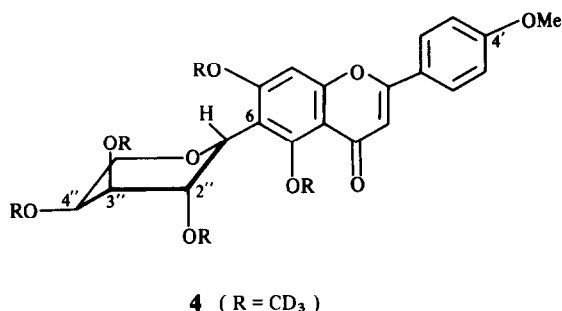
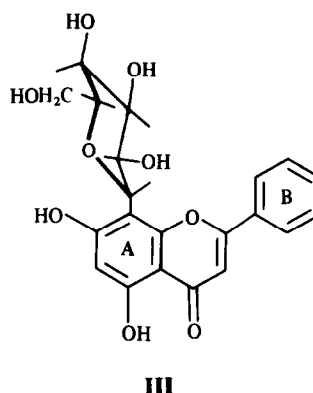
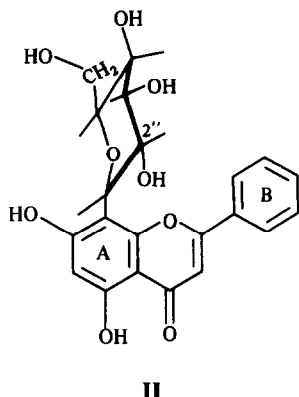


Fig. 3. Effect of positional isomerism and derivatization upon the CD of 6,8-di-C-glycosylflavones containing arabinose and glucose CD spectra of neoschaftoside (10) (—), PDM-10 (---), neoisoschaftoside (11) (—·—), PM-11 (···).

glycosylflavones exist primarily in two principal rotational conformers with the plane of the sugar ring approximately perpendicular to the benzopyrone moiety, II and III (as illustrated for 8-C-glucosylflavones). Generally, the conformation having C-2" above the benzopyrone group, II, has been favored. Thus, Eade *et al.* proposed II as the preferred conformation for bayin hexaacetate on the basis of  $^1H$  NMR spectra [9] and Jurnak and Templeton established that a 6-bromo-hexaacetylated derivative of vitexin possessed a similar conformation in the solid state with the plane of the pyranosyl ring approximately perpendicular ( $86.5^\circ$ ) to the benzopyrone structure [10]. Conformer II, with the pyranose synclinal to the A-ring ( $\theta \sim 45^\circ$ ), also has been proposed by Gentili and Horowitz to account for shielding of the 2"-O-acetyl group by ring A and the 6"-O-acetyl group by ring B in acetylated 8-C-glucosylflavones [11]. More recently,  $^{13}C$  NMR spectra of selected 6-C-glucosylflavones have suggested the presence of a slow conformational equilibrium where interconversion is rapid at  $80^\circ$  and slow at  $30^\circ$  [12].

Factors important in determining the  $^4C_1 \leftrightarrow ^1C_4$  equilibrium\* of the  $\beta$ -pyranosyl ring conformation are the conformational free energy of the R group (flavone) at C-1" and the gauche interaction between the C-2" substituent and the flavone group. Due to the absence of the bulky hydroxymethyl group in aldopentopyranoses, such as arabinose, these sugars often exist in solution as equilibrium mixtures of  $^4C_1$  and  $^1C_4$  conformers [13]. The  $^1C_4$  conformation appears energetically favored for C- $\beta$ -L-arabinopyranosylflavones due to the equatorial disposition of the flavone and 4"-hydroxyl groups (cf. 4).



Unfavorable interactions between either the 5- or 7-substituent on the A-ring and the hydrogens on C-3'' and C-5'' would further destabilize the <sup>4</sup>C<sub>1</sub> conformation.

#### Mono-C-arabinosylflavones

CD spectra of  $\alpha$ -pyranosyl,  $\alpha$ - and  $\beta$ -furanosyl containing C-arabinosylflavones 1–3, 6 and 7 and their derivatives may be interpreted with the aid of the benzoyl sector diagram, I [4]. C-Glycosylflavones 1, 2, 6 and 7 may be assumed to exist in a preferred conformation having the C-2'' atoms above the plane of the A-ring (cf. II in which the pyranose ring or C-2'' of the furanose is synclinal to the A-ring with  $\theta \sim 45^\circ$ ), while 3 is predicted to have the C-2'' atom synclinal below the A-ring, as in III, at an angle of  $\sim -45^\circ$ . Replacement of  $\alpha$ -L-arabinopyranose with  $\alpha$ -D-arabinopyranose at C-6 of a flavone, as in 5, appeared to alter the preferred conformation of the sugar about the C-6, C-1'' bond to one in which the pyranose ring (or C-2'' atom) was at an angle of  $-45^\circ$  (synclinal) to the A-ring. The sign of the charge-transfer band will reflect whichever sector is occupied by C-2'' and its substituent. Derivatization of C-glycosylflavones will alter both the conformation of the carbohydrate about the C-aryl, C-1'' bond and the contribution of perturbing atoms to the magnitude of the CD. The low magnitude of the 250–275 nm CD band shown by  $\beta$ -pyranosylflavone 4 may be due to the presence of nearly equal amounts of conformers II and III (or to conformations having the sugar perpendicular to the A-ring) resulting in cancelled contributions to the CD.

#### Schaftoside and its isomers

The conformational preference about each C-aryl, C-1''

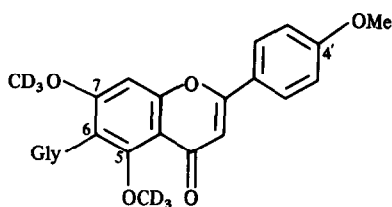
bond in the highly congested permethylated di-C-glycosylflavones is complicated due to the presence of a second sugar in the molecule in addition to 5,7-dimethoxy substituents which may be forced out of the A-ring plane resulting in additional sector contributions. <sup>1</sup>H NMR spectra of PDM-8 and PDM-9 exhibited two doublets for the anomeric proton of each arabinose and glucose residue suggesting two conformers which interconvert very slowly or not at all [8]. The cancelled CD bands at 250–275 nm of PDM-9 may reflect the presence of each sugar in nearly equal amounts of conformers II and III or in orientations perpendicular to the A-ring. On the other hand, a slight preponderance of the 6-C-glucosyl conformer having C-2'' synclinal above the plane of the A-ring ( $\theta \sim 45^\circ$ ) could produce the weak positive band at 261 nm in the CD of PDM-8.

As a result of evidence from <sup>1</sup>H NMR spectra which showed the existence of PDM-10 in two rotameric conformations about the C-aryl, C-1'' bonds in a ratio of 2:1 [8], the strong CD band at 262 nm could result from a preponderance of conformation II with the C-2'' substituent of the glucose group projecting ( $\theta \sim 45^\circ$ ) into the upper left hand sector of diagram I. The lower magnitude of the same CD band in the underivatized 10 may be due to a more equal ratio of conformations II and III, a lessened contribution to the CD of the substituent ( $-\text{OH}$  vs.  $-\text{OCD}_3$ ) or a different angle of twist of the glucosyl group about the C-6, C-1'' bond. The intense negative 271 nm band, which dominated the charge-transfer CD spectrum of 11, suggested the presence of the 8-C-glucosyl residue in a preferred conformation having the C-2'' group synclinal above the plane of the A-ring projected into the upper right sector of I.

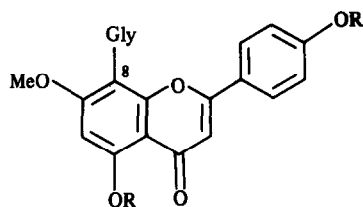
#### EXPERIMENTAL

The CD data recorded in Table 1 were measured in MeOH with the aid of a Cary 60 spectropolarimeter equipped with a 6003 circular dichroism accessory calibrated with (+)-10-camphorsulfonic acid as previously reported [14]. Operating procedures were performed as described earlier [4].

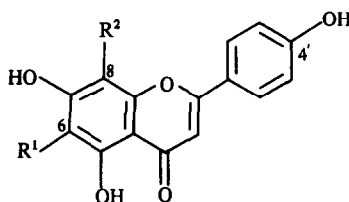
Neoschaftoside (10) was isolated from the liverwort *Radula complanata* by Dr. R. Mues (Saarbrücken) and neoisoschaftoside (11) was isolated from *Gemmungia chinensis* by Dr. K. Yagshita (Yamaguchi). PDM-9, PM-11 and PDM-12 were prepared [Besson, E. and Chopin, J., unpublished results] according to the procedure of Brimacombe, *et al.* [15]. Synthesis of 5 [Dubois,



- 1 Gly = PDM -  $\alpha$  - L - arapyr,  
 2 Gly = PDM -  $\alpha$  - L - arafur,  
 3 Gly = PDM -  $\beta$  - L - arafur,  
 4 Gly = PDM -  $\beta$  - L - arapyr,  
 5 Gly =  $\alpha$  - D - arapyr, (5,7,4' - triOH)



- 6 Gly =  $\alpha$  - L - arapyr, R = H  
 7 Gly = PDM -  $\alpha$  - L - arafur, R = CD<sub>3</sub>



- 8 R<sup>1</sup> = Glc, R<sup>2</sup> =  $\alpha$  - L - arapyr  
 9 R<sup>1</sup> =  $\alpha$  - L - arapyr, R<sup>2</sup> = Glc  
 10 R<sup>1</sup> = Glc, R<sup>2</sup> =  $\beta$  - L - arapyr  
 11 R<sup>1</sup> =  $\beta$  - L - arapyr, R<sup>2</sup> = Glc

- 12 R<sup>1</sup> =  $\beta$  - L - arafur, R<sup>2</sup> = Glc (4' - OMe)  
 13 R<sup>1</sup> =  $\alpha$  - L - arapyr, R<sup>2</sup> = Gal  
 14 R<sup>1</sup> = Gal, R<sup>2</sup> =  $\alpha$  - L - arapyr  
 15 R<sup>1</sup> = Xyl, R<sup>2</sup> =  $\alpha$  - L - arapyr

M. A. and Chopin, J., unpublished results] was achieved by the procedure of Besson and Chopin [2].

The remaining compounds studied in this investigation were described in the following references. 1-4 and 7 [2]; 8 and 9 [3]; PDM-8 and PDM-10 [8], 6 and PDM-6 [16]; 13 [17]; 14 [18] and 15 [19]

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